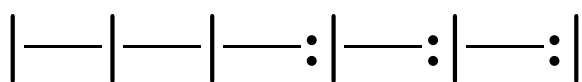


Soviet-era science, translated into English



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Abstract

Full Text

ERRATUM

In our article (V. I. Gusynin, V. L. Tal' roze, "Investigation of energy transfer along the chain $\sim \text{CH}_2 \sim$ by the luminescence-quenching method"), published in *DAN*, vol. 135, no. 5, 1960, Table 2 should read:

Table 2*

Quencher	Luminescing solution	z , l/mol	y , l/mol	x , l/mol
Methyl alcohol CH_3OH	Terphenyl in dioxane	115 ± 12	0.66 ± 0.09	-0.09 ± 0.04
Ethyl alcohol $\text{C}_2\text{H}_5\text{OH}$	Terphenyl in dioxane	115 ± 12	0.93 ± 0.15	-0.06 ± 0.04
Propyl alcohol $\text{C}_3\text{H}_7\text{OH}$	Terphenyl in dioxane	115 ± 12	1.18 ± 0.16	—
Hexyl alcohol $\text{C}_6\text{H}_{13}\text{OH}$	Terphenyl in dioxane	115 ± 12	1.82 ± 0.55	—
Nonyl alcohol $\text{C}_9\text{H}_{19}\text{OH}$	Terphenyl in dioxane	115 ± 12	2.88 ± 0.42	—
Water	Terphenyl in dioxane	115 ± 12	0.23 ± 0.03	—
Carbon tetrachloride	Terphenyl in dioxane	62	28	18
The same	Terphenyl in xylene	415	217	10

* For comparison, values are given that characterize the quenching by water and carbon tetrachloride of the luminescence of a terphenyl solution in dioxane (our own measurements) and in xylene (results taken from work ⁽⁸⁾).

V. I. Gusynin
V. L. Tal' roze

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Note: Figure translations are in progress. See original paper for figures.

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