

Machine Learning the Apparent Diffusion Coefficient of Se(IV) in Compacted Bentonite

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Abstract

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Full Text

Preamble

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Application of Machine Learning in Predicting the Apparent Diffusion Coefficient of Se(IV) in Compacted Bentonite

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Abstract

Light Gradient Boosting Machine (LightGBM) and Random Forest (RF) algorithms were employed to predict the apparent diffusion coefficient of Se(IV) in

compacted bentonite. Seven instances of Se(IV) diffusion were measured using the through-diffusion method. LightGBM ($R^2 = 0.98$ and $RMSE = 0.025$) exhibited superior predictive accuracy when trained on a dataset comprising 956 instances and eight input features sourced from the Japan Atomic Energy Agency (JAEA-DDB) and relevant literature. Shapley Additive Explanation and Partial Dependence Plot analyses provided valuable insights into the diffusion mechanism of adsorbed anions by evaluating the relationships between the apparent diffusion coefficient and each input feature.

Keywords: Diffusion coefficient • Bentonite • Machine learning • Through-diffusion experiment

Introduction

Bentonite is widely utilized as an engineering barrier in high-level radioactive waste repositories to impede radionuclide release. Radionuclide transport in compacted bentonite is governed by diffusion-based mass transport following Fick's law [1–3]. The retardation mechanism is controlled by both diffusion and sorption processes [4]. The apparent diffusion coefficient (D_a) and effective diffusion coefficient (D_e) are two crucial parameters for characterizing Fickian diffusion [5]. The effective diffusion coefficient is calculated using Fick's first law by analyzing radionuclide diffusion at steady-state. Since reaching steady-state requires considerable time, the through-diffusion method is often employed for weakly and non-sorbing radionuclides such as $^{75}\text{Se(IV)}$ [6], HTO [3, 7–9], and EuIII-EDTA^- [10]. In contrast, the apparent diffusion coefficient measured at the transient stage accounts for the accessible porosity of bentonite in the accumulation term of Fick's second law for adsorbed radionuclides [4, 5, 11, 12]. The in-diffusion method is commonly used for strongly sorbing radionuclides such as $^{233}\text{U(VI)}$ [13], $^{237}\text{Np(V)}$ [14], and $^{134}\text{Cs}^+$ [15]. The relationship between the apparent diffusion coefficient (D_a) and the effective diffusion coefficient (D_e) can be expressed as follows [3]:

$$D_a = \frac{D_e}{\varepsilon + \rho_d K_d} = \frac{D_e}{\alpha}$$

where ε , ρ_d , K_d , and α represent porosity, compacted dry density, distribution coefficient, and rock capacity factor, respectively.

Diffusion experiments are generally time-consuming and expensive for obtaining diffusion coefficients. The effective diffusion coefficient is frequently used in diffusion models, including empirical equations and numerical models, enabling rapid and cost-effective predictions. These models can quantitatively describe the dependence of the effective diffusion coefficient of radionuclide species on mineral characteristics (mineral composition, porosity, cation exchange capacity, and external surface area), solution properties (pH, Eh, and mixed ions), and experimental conditions (compacted dry density, temperature, and ionic

strength) [6, 11, 12, 16, 17]. However, the predictive accuracy of these models remains unsatisfactory, likely due to approximations and assumptions made during modeling. For example, the effective diffusion coefficient predicted by Archie's equation deviated from experimental values by approximately 1–1.5 orders of magnitude, primarily because of these approximations [18]. Numerical models such as the integrated sorption and diffusion model [19], multi-porosity model [20, 21], and pore-scale model [22, 23] rarely provide predictive accuracy indices, suggesting unsatisfactory performance. Since in-diffusion experiments for strongly sorbing radionuclides are more complex than through-diffusion experiments [3, 14, 15], limited investigations have measured the apparent diffusion coefficient, which represents the coupled effect of radionuclide sorption and diffusion. Consequently, information regarding its relationship with the aforementioned influencing factors remains scarce.

Machine learning algorithms have been extensively developed for regression prediction [20, 24, 25]. These algorithms employ global analysis techniques such as Shapley Additive Explanation, Individual Conditional Expectation, and Partial Dependence Plots to visually analyze the nonlinear relationships between ion diffusion coefficients and influencing factors [24, 25]. Predictive accuracy depends on the dataset, including input features (influencing factors) and data size. Previous radionuclide diffusion experiments primarily focused on specific experimental conditions, such as radionuclide species, compacted dry density, ionic strength, pH, temperature, and bentonite origin [3, 7–10, 17, 26, 27]. However, most investigations considered only a limited number of influencing factors. The dataset size decreased as the number of input features increased, posing challenges for machine learning applications. This study examined the influence of input features and data size on predicting the apparent diffusion coefficient of Se(IV) in compacted bentonite using the through-diffusion method and two machine learning algorithms: Light Gradient Boosting Machine (LightGBM) and Random Forest (RF), based on a comprehensive diffusion dataset. Furthermore, the diffusion mechanism was investigated by analyzing the weight of influencing factors and the relationship between the apparent diffusion coefficient and each factor.

Materials and Methods

Materials

Gaomiaozi (GMZ) bentonite, obtained from the Beijing Research Institute of Uranium Geology, originates from the Gaomiaozi Mine in Xinghe County (Inner Mongolia, China). It was converted to Ba-bentonite by mixing BaCl_2 with GMZ powder at a mass ratio of 5%. The detailed preparation procedure for Ba-bentonite followed a previous study [28]. The bentonite was compacted into blocks measuring $\varnothing 25.6 \times 12.6$ mm with compacted dry densities of 1300–1700 kg/m^3 .

The Se(IV) stock solution was prepared by dissolving SeO_2 powder (Sinopharm

Reagent) in 0.1–0.5 mol/L NaCl solution. The initial Se(IV) concentration in diffusion experiments was $(12.5 \pm 1.0) \times 10^{-3}$ mol/L for Ba-bentonite experiments and $(17.9 \pm 0.8) \times 10^{-3}$ mol/L for GMZ experiments. All experimental solutions were prepared with ultrapure water from a Milli-Q system (Millipore, USA). Se concentrations were determined using an Optima 7000DV inductively coupled plasma optical emission spectrometer (PerkinElmer, USA).

Diffusion Experiments

Through-diffusion experiments were conducted under ambient conditions at room temperature (22 ± 3 °C). Bentonite blocks were fully saturated in 0.1–0.5 mol/L NaCl solution for 5 weeks. One side of the diffusion cell ($x = 0$) was connected to a source reservoir containing 200 mL of Se(IV) solution, while the other side ($x = L$) was connected to a target reservoir with 10 mL of NaCl solution. Se(IV) diffused through the 12.6 mm thick bentonite block and reached the target reservoir, which was regularly exchanged with fresh 10 mL NaCl solution without Se(IV) to maintain a constant concentration gradient.

Diffusion Data Processing

Two sets of experimental data were obtained: accumulated mass (Acum) of Se(IV) versus time, and flux ($J(L,t)$) versus time [3, 22]. The first dataset was used with the analytical solution of Fick's law to calculate the effective diffusion coefficient (D_e) and rock capacity factor (α) as follows [3]:

$$\text{Acum} = S \cdot L \cdot C_0 \left[\frac{D_e \cdot t}{L^2 \cdot \alpha} - \frac{2 \cdot \alpha}{L^2 \cdot \pi^2} \sum_{n=1}^{\infty} \frac{(-1)^n}{n^2} \exp \left(-\frac{D_e \cdot n^2 \cdot \pi^2 \cdot t}{L^2 \cdot \alpha} \right) \right]$$

where S , L , and C_0 represent the cross-sectional area of the bentonite block, block thickness, and initial Se(IV) concentration in the source reservoir, respectively.

The second dataset was used to verify the effective diffusion coefficient and rock capacity factor as follows:

$$J(L, t) = \frac{\partial \text{Acum}}{\partial t}$$

Machine Learning Database Description and Analysis

The training dataset was sourced from the diffusion database system of the Japan Atomic Energy Agency (JAEA-DDB) (1982–2009) and relevant literature on radionuclide diffusion in bentonite (2010–2023) [8, 9, 21, 22, 28–32]. Insufficient descriptions of influencing factors in JAEA-DDB and the literature reduced the number of instances as the number of influencing factors increased, significantly affecting the predictive performance of ML models.

This study examined the number of instances and input features. Specifically, the input features numbered three, five, and eight, corresponding to 850, 820, and 739 instances in JAEA-DDB (designated J3, J5, and J8). Additionally, 106 instances were collected from published literature, resulting in total instances of 956, 926, and 845 for M3, M5, and M8, respectively. The dataset IM8 contained 956 instances that imputed M8 using the missForest imputation method [33].

The statistical information regarding the number of input features and instances is summarized in Table 1. The input features included compacted dry density, ion diffusion coefficient in water, ionic strength, rock capacity factor, distribution coefficient, montmorillonite content, ionic charge, and temperature. Table S1 in the supporting information provides a summary of montmorillonite content for various bentonite types. Since no literature value for montmorillonite content of Kunipia-P was available, it was assumed equivalent to that of Na-montmorillonite [34]. The output variable was the apparent diffusion coefficient. Both the ion diffusion coefficient in water and apparent diffusion coefficient were transformed into logarithmic form to enhance predictive accuracy.

To assess the predictive performance of machine learning models, root mean square error (RMSE) and coefficient of determination (R^2) were utilized. These metrics were calculated using the following equations:

$$\text{RMSE} = \sqrt{\frac{1}{N} \sum_{i=1}^N (\text{Log} D_{a,i}^{\text{exp}} - \text{Log} D_{a,i}^{\text{pred}})^2}$$

$$R^2 = 1 - \frac{\sum_{i=1}^N (\text{Log} D_{a,i}^{\text{exp}} - \text{Log} D_{a,i}^{\text{pred}})^2}{\sum_{i=1}^N (\text{Log} D_{a,i}^{\text{exp}} - \text{Log} D_a^{\text{exp,ave}})^2}$$

where N represents the number of samples, $\text{Log} D_{a,i}^{\text{exp}}$ and $\text{Log} D_{a,i}^{\text{pred}}$ are the logarithms of the experimental and predicted apparent diffusion coefficients, respectively, and $\text{Log} D_a^{\text{exp,ave}}$ is the average experimental apparent diffusion coefficient. Models demonstrating high accuracy and performance are characterized by low RMSE and high R^2 .

Feature Selection

Feature selection involves identifying the most significant features within a dataset to enhance machine learning model performance. Feature importance measured by the RF algorithm is illustrated in Fig. 1 [Figure 1: see original paper]. The montmorillonite content and ionic strength contributed approximately 5% and 1%, respectively, and were considered weak predictors for this dataset. Based on these results, only six features (rock capacity factor, compacted dry density, ionic charge, distribution coefficient, ion diffusion coefficient in water, and temperature) were selected as D_a predictors for IM6 (containing 956 instances).

Results and Discussion

Measurement of the Apparent Diffusion Coefficient by Through-Diffusion Experiments

The through-diffusion method has been extensively used to study anionic radionuclide diffusion, attributed to high diffusivity resulting from the anionic exclusion effect [9, 10, 22]. ^{79}Se is a crucial radionuclide in repository safety assessment, with Se(IV) primarily present as HSeO_3^- at pH 4–7 [35]. Figure 2 [Figure 2: see original paper] illustrates the breakthrough curves of HSeO_3^- in Ba-bentonite and GMZ bentonite. The accumulated mass of HSeO_3^- exhibited a linear increase during the steady-state stage, which is related to the effective diffusion coefficient. In contrast, the flux showed a significant increase during the transient stage, which is linked to the rock capacity factor. The time required to reach steady-state increased from approximately 10 days at a compacted dry density of 1300 kg/m^3 to about 20 days at 1700 kg/m^3 .

Table 2 lists the diffusion parameters of HSeO_3^- in compacted Ba-bentonite and GMZ bentonite. The apparent diffusion coefficient of Ba-bentonite decreased from 6.8×10^{-11} to $2.2 \times 10^{-11} \text{ m}^2/\text{s}$ as the compacted dry density increased from 1300 to 1700 kg/m^3 . Additionally, the apparent diffusion coefficient of GMZ bentonite increased from 5.9×10^{-11} to $6.7 \times 10^{-11} \text{ m}^2/\text{s}$ as the ionic strength increased from 0.1 to 0.3 mol/L . This value was slightly lower than that reported in synthetic pore water ($D_a = 7.8 \times 10^{-11} \text{ m}^2/\text{s}$) in a previous study [36], which could be attributed to higher pH and greater ionic complexity in the synthetic pore water.

Prediction of Machine Learning Algorithms

Both LightGBM and RF are popular and powerful tree-based machine learning algorithms for regression analysis. They have been applied in various diffusion studies, such as predicting the effective diffusion coefficient of Re(VII) in compacted bentonite [20] and the chloride diffusion coefficient in cement [25]. In this study, they were employed to predict the apparent diffusion coefficient of HSeO_3^- in compacted bentonite. Hyperparameters are values specified before training machine learning models, and optimal performance can be achieved through successful hyperparameter optimization. Table 3 lists the optimized hyperparameters for LightGBM and RF algorithms.

The predictive results for the apparent diffusion coefficient using LightGBM and RF for different training sets are shown in Fig. 3 [Figure 3: see original paper] (J3, J5, J8, M3, M5, M8, IM8, and IM6). Figures 3A–H show LightGBM predictions, while Figs. 3I–P show RF predictions. Figures 3Q and S represent the application of IM8 in predicting D_a values for ReO_4^- , HCrO_4^- , I^- , CeEDTA^- , HTO, and Cs^+ in compacted bentonite [37–40]. These species were selected to represent monovalent radionuclide anions, neutral molecules, and monovalent cations. Notably, the M-models exhibited lower RMSE and higher R^2 compared to the J-models, indicating that a larger number of instances enhanced

predictive performance.

The RMSE ranked in the following order: $J3 > J5 > J8$ and $M3 > M5 > M8$, while R^2 ranked in the opposite order: $J3 < J5 < J8$ and $M3 < M5 < M8$. The predictive accuracy of IM8 was similar to M8 for the LightGBM algorithm, while IM8 exhibited higher predictive accuracy than M8 for the RF algorithm. These observations indicate that an increased number of input features enhanced predictive performance [41].

For J3 and M3, both LightGBM and RF produced relatively low prediction accuracy with R^2 below 0.8, indicating that three input features were insufficient for accurate predictions of the apparent diffusion coefficient of HSeO_3^- in compacted bentonite. However, when the number of input features increased to five, the predictive accuracy of M5 using LightGBM significantly improved, with R^2 of 0.939 and RMSE of 0.042. IM8 demonstrated superior predictive accuracy for both LightGBM ($R^2 = 0.98$ and $\text{RMSE} = 0.025$) and RF ($R^2 = 0.95$ and $\text{RMSE} = 0.039$). The generalization of IM8 in predicting Da values of radionuclides also exhibited good performance, with R^2 of 0.79 and RMSE of 0.34 for LightGBM and R^2 of 0.63 and RMSE of 0.45 for RF (Fig. 3Q and S). This indicates that LightGBM has better generalization ability than RF. LightGBM utilizes gradient-based one-sided sampling, exclusive feature bundling, and histogram-based algorithm techniques, effectively preventing overfitting, accelerating training speed, and reducing memory consumption, thereby improving predictive accuracy. It is worth mentioning that eight input features may be insufficient to fully describe the complexities of radionuclide diffusion in compacted bentonite. Incorporating techniques such as recursive feature elimination or forward selection could further improve model performance and reduce overfitting when additional input features are considered.

The predictive accuracy (R^2 and RMSE) was similar for both IM8 and IM6 (Fig. 3G, H, O, and P). However, application tests showed that IM6 had lower predictive accuracy than IM8 (Fig. 3Q–T). Therefore, only the IM8 model is considered in this study.

Spearman and Shapley Additive Explanation Analyses

Figure 4 [Figure 4: see original paper] shows the correlation of each feature with Da in the training dataset using Spearman analysis and the weight of input features for IM8 using Shapley Additive Explanation (SHAP) analysis. Models J3–J8 and M3–IM8 showed similar Spearman correlation coefficients, indicating that the number of input features and instances had insignificant impact on the nonlinear relationship (Fig. 4A). Four parameters (compacted dry density, rock capacity factor, distribution coefficient, and ionic charge) were negatively correlated with the apparent diffusion coefficient, while three parameters (ion diffusion coefficient in water, montmorillonite content, and temperature) exhibited positive correlation. Ionic strength had an insignificant impact. The rock capacity factor, distribution coefficient, compacted dry density, and ionic charge

had high absolute Spearman correlation coefficient values, indicating strong correlations with the apparent diffusion coefficient.

The Shapley Additive Explanations (SHAP) method is commonly used to analyze the weight or importance of input features on predictions. SHAP results for LightGBM and RF demonstrated approximately similar ranking orders (Fig. 4B). Specifically, LightGBM ranked the ion diffusion coefficient in water fifth, whereas RF ranked it fourth. The top two input features for predicting the apparent diffusion coefficient were the rock capacity factor and compacted dry density, contributing a total of 53.3% for LightGBM and 53.7% for RF compared to all features.

Montmorillonite content and ionic strength are two important parameters that have been extensively investigated, as they significantly influence the effective diffusion coefficient of anionic radionuclides [8, 22, 42]. However, SHAP analysis demonstrates that montmorillonite content and ionic strength made insignificant contributions to the prediction (Fig. 4B). The tendency for ionic strength is consistent with [9], who reported that its insignificant influence on the apparent diffusion coefficient of Cs^+ and Na^+ was attributed to the coupled effects of diffusion and sorption. Further studies are needed to verify these results, as the weight of input features is influenced by different types and numbers of instances, as well as machine learning algorithms.

Partial Dependence Plots Analysis

Partial Dependence Plots (PDP) can visualize the relationship between each input feature and the predicted apparent diffusion coefficient. Figure 5 [Figure 5: see original paper] shows the contribution of each input feature to the prediction for IM8 using PDP analysis. The gray column shows the distribution of data points for an input feature. PDP values for LightGBM are represented by solid lines, while dashed lines represent RF values. A flat curve suggests that the feature has an insignificant influence on the predicted outcome, whereas a steep curve indicates a stronger relationship. Both LightGBM and RF demonstrate similar trends for the dependency of the predicted apparent diffusion coefficient on each input feature.

The analysis of the relationship between the predicted apparent diffusion coefficient and input features related to experimental conditions was consistent with published experimental results [3, 13, 26]. Specifically, it showed a negative correlation with compacted dry density (Fig. 5A). This can be explained by the decrease in available pores for radionuclide diffusion with increasing compaction. At compacted dry densities above 1600 kg/m^3 , the decreasing tendency changed slowly, attributed to the reduction of interlayer pores to a single water layer [21, 42, 43].

The apparent diffusion coefficient exhibited a positive correlation with temperature (Fig. 5B). In a repository, the temperature of clay near waste containers could exceed 100°C [44]. The positive relationship between the effective dif-

fusion coefficient and temperature was reported for HTO, $^{36}\text{Cl}^-$, and ReO_4^- , following the Arrhenius equation [3, 7, 26]. Figures 5C and D show that the apparent diffusion coefficient exhibited insignificant correlations with montmorillonite content and ionic strength. Slight negative and positive impacts were observed for montmorillonite content ranging from 0.5 to 0.8 and ionic strength below 1.0, respectively, which agrees with their correlations with the effective diffusion coefficient [20]. Numerous studies have shown that as ionic strength increases, the effective diffusion coefficient of radionuclides also increases [8, 22]. This tendency was explained by the decreased thickness of the electrical double layer in high-salinity solutions [9, 22]. In general, the consistency between PDP and experimental results implies that PDP analysis can provide insights into the diffusion laws and mechanisms of radionuclides.

The input features—rock capacity factor, distribution coefficient, ion diffusion coefficient in water, and ionic charge—are parameters related to radionuclide properties. Their dependencies on the apparent diffusion coefficient were unclear due to the coupled effects of solid, liquid, and radionuclide interactions. Figure 5E indicates that the rock capacity factor had a negative impact on the prediction. According to Eq. (3), the apparent diffusion coefficient is inversely proportional to the rock capacity factor. The negative relationship between the apparent diffusion coefficient and distribution coefficient can be attributed to the linear correlation between the rock capacity factor and distribution coefficient (Fig. 5F). The predicted apparent diffusion coefficient increased with increasing ion diffusion coefficient in water (Fig. 5G), indicating that radionuclides diffuse quickly in both liquid and solid phases. The RF algorithm shows that ionic charge had an insignificant effect on the apparent diffusion coefficient, whereas the LightGBM algorithm demonstrates that it decreased as the ionic charge reduced from 0 to +3 (Fig. 5H).

Conclusions

The effects of input features and instances on predicting the apparent diffusion coefficient were investigated using Light Gradient Boosting Machine (LightGBM) and Random Forest (RF) algorithms. HSeO_3^- diffusion experiments in compacted bentonite were conducted using a through-diffusion method to validate predictive performance. Increasing the number of input features resulted in a decrease in instances. LightGBM ($R^2 = 0.98$ and $\text{RMSE} = 0.025$) and RF ($R^2 = 0.95$ and $\text{RMSE} = 0.039$) exhibited superior predictive accuracy for the training set of 956 instances and eight input features: compacted dry density, ion diffusion coefficient in water, ionic strength, rock capacity factor, distribution coefficient, montmorillonite content, ionic charge, and temperature.

Shapley Additive Explanations revealed that the top two input features for predicting the apparent diffusion coefficient were the rock capacity factor and compacted dry density. Partial Dependence Plots indicated that the apparent diffusion coefficient was negatively dependent on the rock capacity factor, compacted dry density, and distribution coefficient, while a positive relationship

was observed between the apparent diffusion coefficient and the ion diffusion coefficient in water. This study presents a method for predicting the apparent diffusion coefficient, quantifying influencing factors, and understanding the effect of each input feature on the apparent diffusion coefficient. The insights into diffusion mechanisms are beneficial for repository safety assessment.

Supplementary Information The online version contains supplementary material available at <https://doi.org/10.1007/s10967-024-09637-w>.

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