

Effects of B Doping on the Structure and Magnetic Properties of SmFe₁₀Mo₂ Alloy Postprint

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Abstract

Master alloy ingots of SmFe₁₀Mo₂ and SmFe₁₀Mo_{1.5}B_{0.5} were prepared by arc melting. After homogenization annealing, SmFe₁₀Mo_{1.5}B_{0.5} nanocrystalline alloy powder was fabricated by ball milling. The effects of B doping on the phase structure and magnetic properties of SmFe₁₀Mo₂ bulk alloy and the influence of ball milling on SmFe₁₀Mo_{1.5}B_{0.5} nanocrystalline powder were investigated. The results show that after B doping, the ThMn₁₂ phase structure of the alloy remains unchanged, while the Curie temperature increases from 270°C to 334°C; the compositional inhomogeneity of the alloy leads to two phase transition points in the thermomagnetic curve. After ball milling for 0.5 h, Mo precipitates significantly and the 1:12 phase decreases markedly in the SmFe₁₀Mo_{1.5}B_{0.5} alloy; with increasing ball milling time, α -Fe precipitates and forms a non-magnetic Mo₂FeB₂ phase, causing a significant decrease in intrinsic coercivity, and its saturation magnetization shows a trend of first increasing and then decreasing with ball milling time. The nanocrystalline alloy powder ball-milled for 0.5 h exhibits the optimal magnetic properties: $M_s = 55 \text{ A} \cdot \text{m}^2/\text{kg}$, $iH_c = 0.2 \text{ T}$.

Full Text

Preamble

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Effect of B-doping on the Structure and Magnetic Properties of SmFe₁₀Mo₂ Alloy

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Abstract

Ingots of SmFe₁₀Mo₂ and SmFe₁₀Mo_{1.5}B_{0.5} master alloys were prepared by arc melting, homogenized through annealing, and then used to produce SmFe₁₀Mo_{1.5}B_{0.5} nanocrystalline alloy powders via ball milling. The effects of B doping on the phase structure and magnetic properties of bulk SmFe₁₀Mo₂ alloy, as well as the influence of ball milling on the nanocrystalline SmFe₁₀Mo_{1.5}B_{0.5} powder, were investigated. The results show that B doping does not alter the ThMn₁₂-type phase structure of the alloy, but significantly increases the Curie temperature from 270°C to 334°C. Compositional inhomogeneity in the alloy leads to the appearance of two phase transition points in the thermomagnetic curve. After ball milling for 0.5 h, substantial Mo precipitation occurs in the SmFe₁₀Mo_{1.5}B_{0.5} alloy, accompanied by a marked reduction in the 1:12 phase. With increasing milling time, α -Fe precipitates and a non-magnetic Mo₂FeB₂ phase forms, causing a significant decrease in intrinsic coercivity. The saturation magnetization initially increases and then decreases with prolonged milling time. The nanocrystalline alloy powder milled for 0.5 h exhibits optimal permanent magnetic properties: $M_s = 55 \text{ A} \cdot \text{m}^2/\text{kg}$ and $iH_c = 0.2 \text{ T}$.

Keywords: metallic materials, ball milling, phase structure, permanent magnetic properties, Curie temperature

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Introduction

Rare earth elements can form stable ThMn₁₂-type rare earth compounds R(Fe, T)₁₂ with transition elements such as Fe, Ti, V, and Mo [1]. These rare earth-iron intermetallic compounds have become another excellent permanent magnetic material following Nd₂Fe₁₄B [2-4]. Among them, SmFe₁₁Ti exhibits the best performance with a maximum energy product $(BH)_{\text{max}} = 268 \text{ kJ/m}^3$, remanence $B_r = 1.16 \text{ T}$, and Curie temperature $T_C = 312^\circ\text{C}$, showing promise for practical applications [5]. RFe₁₀Mo₂ compounds readily undergo spin reorientation at low temperatures [6], and SmFe_{12-x}Mo_x has attracted widespread attention due to its high magnetic anisotropy [7]. The crystal structure of this compound is the ThMn₁₂-type symmetric structure with space group I4/mmm, containing two R(Fe, T)₁₂ molecules (26 atoms) per unit cell [8]. Rare earth Sm atoms occupy the 2a crystallographic site, while Fe and Mo atoms can be distributed across three 3d sites (8i, 8j, and 8f), with Mo atoms preferentially occupying the 8i site [7].

The addition of trace interstitial atoms such as H, C, and N to R(Fe, T)₁₂ compounds significantly improves magnetic performance [9]. Partial substitution of T elements in R(Fe, T)₁₂N_x with V, B, Co, Mo, and other elements can also enhance magnetic properties [10]. Studies on Sm(Fe, Mo)₁₂ interstitial nitrides [11] and hydrides [9] have shown that trace N and H atoms occupying interstitial lattice sites enhance the exchange interaction between Fe-Fe atoms and increase the Curie temperature of the compounds. Mo plays a dual role in 1:12 interstitial nitrides: on one hand, high Mo content tends to reduce the saturation magnetization and Curie temperature of R(Fe, T)₁₂N_x; on the other hand, substituting T elements with Mo in R(Fe, T)₁₂N_x can yield higher coercivity [7, 12]. Partial substitution of Fe with Mn leads to a decrease in the Curie temperature of SmFe_{10.5}Mo_{1.5} alloy due to the combined effect of site occupation [13]. Y. B. Kim et al. [14] investigated the changes in intrinsic magnetic properties caused by B substitution for Ti in NdFeTi nitrides, finding that B addition increased the Curie temperature and saturation magnetization of the nitrides by 20% and 15%, respectively. In this work, SmFe₁₀Mo₂ master alloy ingots were prepared by arc melting, and SmFe₁₀Mo_{1.5}B_{0.5} alloy was synthesized. Nanocrystalline SmFe₁₀Mo_{1.5}B_{0.5} alloy powders were then prepared by high-energy ball milling to study the influence of milling time on phase structure and magnetic properties.

Experimental Methods

Rare earth element Sm with purity greater than 99.99%, and metallic Fe and Mo with purity greater than 99.5% were used as starting materials. The elements were weighed according to the desired chemical compositions and melted in a vacuum induction furnace or arc furnace under high-purity argon protection to prepare SmFe₁₀Mo₂ master alloy. FeB alloy (Fe: 80%, B: 20%) was added to produce SmFe₁₀Mo_{1.5}B_{0.5} alloy. Due to the high vapor pressure of rare earth element Sm, an extra 30 wt% of Sm was added during melting. The as-cast alloy ingots exhibited compositional inhomogeneity and required homogenization annealing to form the desired phase structure and reduce α -Fe content. The melted ingots were wrapped in molybdenum foil and homogenized at 950°C for 7 days under high-purity argon protection. The annealed samples were water-quenched, and surface oxide layers were removed for subsequent experiments.

The melted master alloys were coarsely crushed and then sealed in a ball mill jar with steel balls in a high-purity argon atmosphere glove box, with a ball-to-powder weight ratio of 20:1. SmFe₁₀Mo_{1.5}B_{0.5} alloy powders were prepared using a GN-2 high-energy ball mill at 100 V with a rotation speed of 600 r/min.

Phase composition and surface morphology were characterized using a Rigaku D/max-2500 X-ray diffractometer (XRD) (operating at 50 kV and 300 mA with Cu K α radiation ($\lambda = 0.150456$ nm)) and an S-3400N scanning electron microscope (SEM). The Curie temperature of bulk alloys and room-temperature magnetic hysteresis loops of SmFe₁₀Mo_{1.5}B_{0.5} powders were measured using Lakeshore 7407 and VSM-220 vibrating sample magnetometers.

Results and Discussion

2.1 Phase Structure and Morphology Analysis

[Figure 1: see original paper] shows the X-ray diffraction patterns of bulk SmFe₁₀Mo₂ alloy and SmFe₁₀Mo_{1.5}B_{0.5} alloy powders milled for different durations. The master alloy bulk consists primarily of the 1:12 phase with ThMn₁₂ structure. The 1:12 phase structure remains unchanged with B addition. However, Curie temperature measurements indicate that SmFe₁₀Mo_{1.5}B_{0.5} alloy exhibits two phase transition points, attributed to compositional inhomogeneity in the arc-melted bulk alloy. B enters the lattice of some SmFe₁₀Mo₂ compounds, forming Sm(Fe, Mo, B)₁₂ phase that coexists with the SmFe₁₀Mo₂ phase. Our previous research found that ball milling significantly affects the phase composition of Sm₂Co₁₇ alloys [16]. Tang et al. [17] demonstrated that ball milling induces decomposition of the 1:12 phase in NdFe₁₁Ti alloy, thereby improving magnetic properties. To investigate the influence of ball milling on the phase structure and magnetic properties of SmFe₁₀Mo_{1.5}B_{0.5} alloy, experiments with different milling durations were conducted. As shown in the figure, substantial Mo precipitation occurs after 0.5 h of milling. With further increase in milling time, α -Fe precipitation increases, and the main phase transforms to Mo₂FeB₂. After 2 h of milling, α -Fe precipitates extensively and becomes the dominant phase, while the phase composition remains essentially unchanged after 3 h. Meanwhile, diffraction peaks broaden significantly due to reduced grain size and increased internal stress during milling. The grain size calculated using the Scherrer formula is 21.7 nm for the 0.5 h milled sample and decreases to 7.4 nm for the 2 h milled sample.

[Figure 2: see original paper] presents SEM micrographs of SmFe₁₀Mo_{1.5}B_{0.5} alloy powders milled for 0.5 h and 2 h. The particle size distribution is non-uniform, with irregular shapes and typical agglomeration phenomena, similar to observations by T. Tao et al. in ball-milled tin dioxide powders [18]. After 2 h of milling, large particle clusters decrease noticeably, distinct boundaries become visible, and fine particles increase. Some large particles may result from agglomeration of overly fine powders with increased stickiness [8]. The particle sizes are 1–5 μ m and 0.5–5 μ m for samples milled for 0.5 h and 2 h, respectively.

2.2 Curie Temperature and Magnetic Properties

[Figure 3: see original paper] shows the thermomagnetic curves of SmFe₁₀Mo₂ and SmFe₁₀Mo_{1.5}B_{0.5} alloys, where curves (a) and (b) were measured at 0.01 T, and curve (c) at 0.05 T. The Curie temperature is defined as the temperature corresponding to the minimum of the first derivative of magnetization with respect to temperature on the thermomagnetic curve. As seen in Figures 3b and 3c, bulk SmFe₁₀Mo₂ alloy shows only one phase transition point corresponding to the 1:12 phase, whereas both bulk and 0.5 h milled nanocrystalline SmFe₁₀Mo_{1.5}B_{0.5} powders exhibit two phase transition points, corresponding

to different compositions of the 1:12 phase. This result confirms that B-doped bulk alloys prepared by arc melting have compositional inhomogeneity. One transition temperature remains consistent with the master alloy bulk at approximately 270°C, confirming the presence of SmFe₁₀Mo₂ phase, while the other transition corresponding to Sm(Fe, Mo, B)₁₂ phase increases substantially to 334°C, demonstrating that B doping significantly enhances the Curie temperature. This result is similar to the finding by C. S. Kim et al. [19] that B substitution for Ti increased the Curie temperature of NdFe_{10.7}TiB_{0.3} by 17°C. The Curie temperature of R-T compounds depends primarily on T-T exchange interactions, which are strongly influenced by interatomic distances [8]. B substitution for partial Mo alters T-T interatomic distances, enhancing ferromagnetic exchange and thereby increasing the Curie temperature. Notably, compared with the bulk sample, the 0.5 h milled sample shows weakened Curie temperature peaks, with both transition temperatures decreasing from 272°C and 334°C to 250°C and 315°C, respectively. This is mainly attributed to changes in phase composition and reduced particle size induced by ball milling. X. C. Kou et al. [7] found that higher Fe content in SmFe_{12-x}Mo_x leads to higher Curie temperature of the 1:12 phase. As shown in the XRD patterns in Figure 1, extensive Mo precipitation and Mo₂FeB₂ phase formation in the 0.5 h milled sample alter the Fe/Mo composition ratio in the 1:12 phase, resulting in Curie temperature changes.

[Figure 4a: see original paper] shows the demagnetization curves of alloys before and after B doping. B addition decreases the saturation magnetization without significantly improving coercivity. The thermomagnetic curves in Figure 3 indicate that B introduction creates two distinct 1:12 phases in the alloy: SmFe₁₀Mo₂ phase and Sm(Fe, Mo, B)₁₂ phase. Non-uniform B distribution leads to only modest coercivity improvement and reduced saturation magnetization in the bulk alloy. [Figure 4b: see original paper] presents demagnetization curves of samples milled for different durations. Coercivity increases significantly from 0.03 T to 0.2 T after 0.5 h of milling. The coercivity of magnets depends primarily on microstructural features (size, orientation, boundary defects) [20]. Ball milling refines grains and increases grain boundaries, enhancing domain wall pinning at grain boundaries and thereby increasing coercivity [21]. With increasing milling time, the 1:12 phase gradually decreases, α -Fe soft magnetic phase precipitates extensively, and grain refinement occurs. Severe lattice distortion within particles destroys intrinsic crystalline properties, weakening grain boundary pinning effects on domain walls [22], leading to a sharp decrease in coercivity and gradual increase in saturation magnetization of the nanocrystalline powders. After 2 h of milling, α -Fe precipitation saturates, reaching a maximum saturation magnetization of 78 A · m²/kg. Further milling causes extensive Mo incorporation into the Mo₂FeB₂ phase, weakening Mo diffraction peaks, further reducing powder particle size, and promoting agglomeration and oxidation [23], which decreases saturation magnetization to 70 A · m²/kg. The alloy powder milled for 0.5 h exhibits optimal magnetic properties: saturation magnetization $M_s = 55 \text{ A} \cdot \text{m}^2/\text{kg}$ and intrinsic coercivity $iH_c = 0.2 \text{ T}$.

Conclusions

1. B-doped SmFe₁₀Mo₂ alloy maintains the ThMn₁₂-type phase structure, with Curie temperature significantly increased from 270°C to 334°C. Compositional inhomogeneity leads to two phase transition points in the thermomagnetic curve and reduces alloy saturation magnetization.
2. Ball milling of SmFe₁₀Mo_{1.5}B_{0.5} alloy significantly increases both coercivity and saturation magnetization. With increasing milling time, substantial Mo and α -Fe precipitation occurs, the main phase decreases markedly, and a non-magnetic Mo₂FeB₂ phase forms, causing a sharp decrease in nanocrystalline powder coercivity and a saturation magnetization trend that first increases then decreases.

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